Forensic toxicology: importance in crime investigation

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Abstract
By using chemical and analytical techniques, forensic toxicology contributes to the establishment of facts in forensic investigations by studying the medical and legal aspects of the detrimental effects of a drug on organisms. Forensic practises is crucial in cases of deadly poisoning and those that could be connected to criminal activity. In most murder, suicide, or accident cases, poisons are discovered. They are a crucial component of the silent weapon that silently and covertly ends life. The current study is focused on the field of forensic chemistry and toxicology, which is entirely centered on the introduction, categorization, effects, and influencing factors of poisons, as well as their detection and testing. This article’s goal is to examine how they behave and perform after they enter the human body. Poisons have serious effects and might even be fatal if not treated appropriately.

Key words: Forensic, Toxicology, Poisoning, Neurotoxins, Irritants

1. Introduction

The word "forensic" comes from the Latin word "forensis," which means "public," "to the prospect and discussion"[1]. Legal concerns can be resolved in a variety of ways thanks to rhetorical science. Rhetorical science, which includes a variety of disciplines such as rhetorical chemistry, rhetorical social science, rhetorical biology, rhetorical medicine, rhetorical engineering, rhetorical material sciences, machine rhetorical, and others, is generally used to settle legal disputes, fairly enforce criminal and civil laws, and protect the public. A broad phrase that covers most of the duties performed by the law laboratory is forensic chemistry[2]. Trace analysis and medicine are techniques employed in the field of rhetorical chemistry. Analytical chemistry that is applied is forensic chemistry. Rhetorical chemistry adds comparison study to the assignment whereas analytical chemistry covers both quantitative and qualitative chemical analysis [3]. Similarly, spectrometry can quickly determine whether a sample is made of nylon or polythene. Therefore, analytical chemistry offers the quantitative and qualitative data needed to respond to rhetorical questions [4].
Forensic Analysis
The use of scientific knowledge that is based on legal issues is known as forensic science [5]. In order to determine what happened, when it happened, and who was responsible, forensic science primarily examines biological and physical evidence. To attain accuracy and precision, forensic scientific proficiency is crucial [6].

2. Forensic Toxicology

The study of poisons' and drugs' detrimental effects on living things is known as toxicology [7]. It include the investigation of the signs, causes, effects, and methods of handling certain toxins and medications. If the use of medications and poisons results in death in dubious circumstances, it becomes rhetorical pharmacological medicine [8].

The field of medicine may include pharmacological medicine as a sub-field. The study of medicine includes all interactions between drugs and other chemicals on living things [9]. Drug administration, body absorption, activities and interactions, metabolism, and excretion are all aspects of medicine [10].

![Figure 9: Stages of forensic toxicology](image)

3. History of Forensic Chemistry

Rhetorical chemistry advancements started to become apparent by the middle of the 19th century. Blood tests were developed at this time, the Marsh test for arsenic was created in 1832, and experiments on bullet "fingerprinting" were conducted in the 1980s. Christian Friedrich Schönbein (1799–1868), a German–Swiss scientist, developed the first accurate method for differentiating human blood in 1863 [11]. For more than a century and a half, arsenic has been widely used as a toxin. Its excellence goes back to the seventh century. During the 16th century, the Arab philosopher Abu Musa Jabir Ibn Hayyan (about 721-ca. 815), also known as Geber, found the process for transforming the grey, metallic-looking elemental arsenic, an elemental chemical compound (As2O3; a material) into (White, flavourless, and odourless powder). Arsenic may just be an extra chemical in person's food or beverage without raising suspicion [12]. Over the following sixty years, attempts to employ bullet "fingerprinting" in criminal investigations were few [13]. But by the 1890s, a number of events spurred renewed interest in the method as a means of identifying possible offenders. Of those aspects, the development of the replacement method known as "grooving" for making gun barrels was perhaps the most crucial. The process of grooving involves carving spiral grooves into the inner surface of tubing. When a bullet travels through the barrel, the grooves allow it to spin; this motion prevents the bullet from tipping over after it exits the tube. Different grooving techniques have been employed by numerous gun makers [14].

Poisoning
It is referred to as the harmful consequence of a poison or harmful chemical agent. It causes the emergence of negative reactions to dangerous substances or chemicals [15]. Basically, it
may be divided into three groups: homicidal, suicidal, and accidental [16].

**Homicidal Poisoning**
The victim is frequently subjected to attempts to "nurse" them back to health by poisoners [17]. Serial poisoners typically enjoy the rush of having control over the victim’s life and suffering, and poisons frequently take delight in seeing their victims suffer. Homicide by poisoning perpetrators frequently work in the healthcare or medical industries. The substances that are most appealing to offenders are those that are deadly in little doses. The ideal poison for a homicide has no taste, is undetectable, has no odour, and exhibits symptoms that are comparable to those of illnesses that are found in nature [18]. Since current scientific techniques and advancements have made poison detection simpler, it has becoming more and more challenging to find a poison having all of these characteristics [19].

**Suicidal Poisoning:**
Self-poisoning, the non-violent way of suicide, most frequently involves the use of medicine, either over-the-counter (paracetamol) or prescribed (such as antidepressants and prescription analgesics), chemicals (pesticides), or illegal narcotics [20].

**Accidental poisoning**
Accidental poisoning, which includes accidental drug overdose, occurs when a person inadvertently poisons themself. Alcohol, opioids (such as heroin or methadone), sedatives, psychiatric pharmaceuticals (such as antidepressants), antiepileptic, and anti-inflammatory medications are some of the substances from which poisoning may result [21].

**Classification of poisons**
Poisons are divided into two categories based on how they affect the body and depending on their chemical and physical characteristics.

**Classification based upon the effect of poison on the body:**

A) **Corrosive:** When poisons come into contact with tissues or organs, they become corrosive, for example: a. Strong acids like H2SO4, HCl, HNO3, etc. and strong alkalis include NH4, Na/K hydroxides, etc [22].

B) **Neurotoxins:** Toxins known as neurotoxins cause damage to nerve tissue. Exogenous chemicals known as neurotoxins are a broad category of neurological insults that can negatively impact the function of both growing and mature brain tissue. Lead, glutamate, ethanol (drinking alcohol), botulinum toxin (e.g., Botox), tetanus toxin, nitric oxide and tetrodotoxin are typical examples of neurotoxins [23].

C) **Irritants poisons:** They mostly cause inflammation at the point of contact, particularly in the skin, gastrointestinal system, and respiratory tract [24]. A poison is categorized as one that affects a system the most when it causes death as a result of a systemic impact, such as a heart poison, brain poison, or spinal poison. The inorganic toxin arsenic is a hefty metallic irritation. Due to its insoluble nature in water and inability to be absorbed by the digestive system, metallic arsenic is not harmful. Arsenic trioxide, often known as sanguinaria or somalkar, is toxic. Arsenobetaine and arsenocholine are two organic arsenic non-toxic forms that are typically present in food that humans frequently ingest [25]. Cod, haddock, and shellfish contain them.
Poisons are also Categorized According to their Characteristics:

A) Inorganic Poisons
i) Metallic Poisoning
After exposure, microscopic metal molecules build up in your body and cause heavy metal poisoning. Without treatment, heavy metals can produce symptoms that are potentially fatal because they adhere to your cells and stop them from functioning. Your body can become poisonous to a variety of metals [26]. The most prevalent poisonous metals are:

<table>
<thead>
<tr>
<th>Types of Metals</th>
<th>Where they can be found</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cadmium</td>
<td>metal plating, Cigarette smoke ,batteries.</td>
</tr>
<tr>
<td>Arensic</td>
<td>Topical creams, polluted water, shellfish, algae, pesticides, fungicides, insecticides, fungicide-based paints, enamels, and glass.</td>
</tr>
<tr>
<td>Lead</td>
<td>lead pipes, batteries, paint, gasoline, and building materials have all tainted the water.</td>
</tr>
<tr>
<td>Mercury</td>
<td>Batteries, shellfish, topical antiseptics, dental amalgam (&quot;silver&quot;) fillings, thermometers, and lightbulbs</td>
</tr>
<tr>
<td>Thallium</td>
<td>Rodenticides, fireworks , pesticides,.</td>
</tr>
</tbody>
</table>

ii) Non Metallic Poisons

A) Phosphine and phosphides
A variety of industrial operations employ phosphonate, a highly poisonous colourless gas with a pungent garlic or fishy odour. It is also produced when phosphides are exposed to moisture.[27] In underdeveloped nations, aluminium phosphide is widely utilized as a cheap and efficient grain fumigant and rodenticide [28].

B) Bromide
The colourless gas methyl bromide has historically been employed as a refrigerant and in fire extinguishers, but it is most frequently utilized as an insecticidal fumigant for grain storage and soil. Methyl bromide is a metabolite that produces the bromide ion and is linked to unintentional poisoning, especially in work environments [29]. By passive diffusion, the bromide ion is quickly absorbed from the stomach and proximal small intestines. Bromide ions are mostly found in extracellular fluid, where they have similar properties to chloride ions. The kidney is the most significant organ for elimination. Their half-life of elimination is relatively lengthy, lasting around 10 days after an acute dose or many weeks after stopping a long-term consumption, especially in situations of bromide intoxication.
C) Cyanides
The majority of cases of severe or deadly cyanide poisoning involve the suicide intake of cyanide salts. A very poisonous volatile liquid, hydrogen cyanide. When cyanide salts react with acids or are formed in the stomach after oral consumption, hydrogen cyanide fumes are released [30]. Although HCN has a distinctive almond-like odour, up to 50% of people cannot detect it. Even while this might have been connected to air flow ventilation systems in post-mortem rooms, it was surprising that it was not a distinguishing feature during autopsy of a significant number of cyanide suicide fatalities. The industrial applications of potassium and sodium cyanide as soluble salts of cyanide include electroplating, metal processing, and laboratory reagents [31].

B. Organic poisons

a) Ethanol: A much of ethanol is toxic, so avoid using it.

b. Other alcohols: Poisonous alcohols include methyl and isopropyl.

Methanol, a substance used in the chemical and polish industries as well as clandestine alcoholic beverages, may be fatal when consumed.

c. Phenol: Carboxylic acid or phenol may be toxic. The main purpose of it is as a disinfectant.

d. Other substances: Poisonous industrial chemicals include benzene, chloral hydrate, chlorinated hydrocarbons, and others. Chloral hydrate might be present in illegal alcoholic beverages in a number of poisoning instances [32].

Route & Site of Exposure
When administered intravenously, toxic substances often have the most impact and the fastest reaction. For alternative routes, inhalation, intraperitoneal, scutaneous, intramuscular, intradermal, oral, and topical would roughly be listed in decreasing order of efficacy [33]. Additionally, the method of delivery might affect an agent's toxicity. For instance, it would be reasonable to anticipate that a substance that is detoxified in the liver would be less hazardous when administered orally through the portal circulation than systemically (inhalation) [34].

Duration & Frequency of Exposure
The toxic effects brought on by a single exposure to multiple substances differ significantly from those brought on by repeated exposure. For instance, benzene's major acute toxic symptom is central nervous system depression, but prolonged exposure can cause leukaemia. Acute exposure to quickly absorbed substances is likely to result in immediate toxicity, but it is also possible for acute exposure to result in delayed toxicity that may or may not be comparable to the toxic consequences of chronic exposure. In contrast, repeated administration of a hazardous agent may result in certain short-term (acute) side effects in addition to the agent's long-term, low-level, or chronic effects [35].

Local versus Systemic Toxicity
The central nervous system is the target organ of toxicity that is most commonly engaged in systemic toxicity. Damage to the central
nervous system, particularly the brain, can be proved by the use of proper and sensitive procedures, even if many substances have obvious effects elsewhere. The circulatory system, the blood and hematopoietic system, visceral organs such as the liver, kidney, and lung, and the skin follow in order of frequency of involvement in systemic toxicity. The least often targeted tissues for systemic effects include muscle and bone. The frequency of tissue reactions when drugs have a localized effect primarily depends on the portal of entrance (skin, gastrointestinal tract, respiratory tract) [36].

**Reversible versus Irreversible Toxic Effects**

Chemical toxicity can have both reversible and irreversible consequences. The tissue's capacity to regenerate will play a significant role in determining whether a chemical insult to a tissue results in reversible or irreversible damage. Therefore, whereas most injuries to the central nervous system are mostly irreversible due to the differentiated cells of the central nervous system being unable to divide and be replaced, most injuries to a tissue like the liver, which has a high capacity for regeneration, are reversible. Chemicals can cause cancer and permanent harmful consequences [37].

**4. Conclusion**

By analyzing the medical and legal elements of a drug's harmful effects on organisms, forensic toxicology uses chemical and analytical tools to help establish the facts in forensic investigations. In situations of fatal poisoning and those that may be related to criminal conduct, forensic practise is essential. Poisons are typically found in murder, suicide, or accident situations. They are an essential part of the silent weapon that kills people stealthily and invisibly. The subject of the current research is forensic chemistry and toxicology, which is fully concerned with the introduction, classification, effects, and influencing factors of poisons, as well as their detection and testing. The purpose of this article is to investigate how they act and function after they enter the human body.

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